

Wet Deposition in SILAM

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Outline:

- ▶ SILAM briefly
- ▶ WD in SILAM old vs. new
- ▶ Mechanistic WD
- ▶ Special case: SO₂
- ▶ Improvements due to new scheme
- ▶ Conclusions



SILAM v.5: outlook

Modules

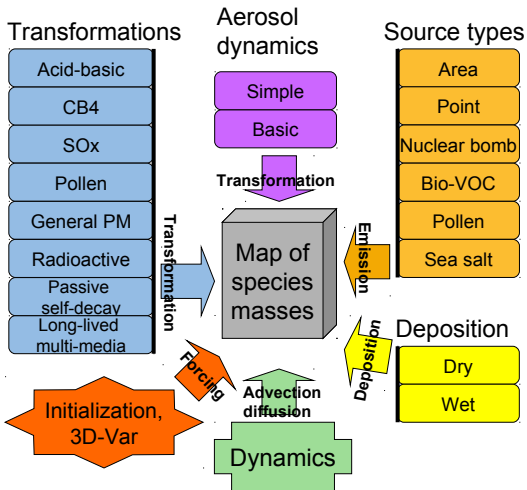
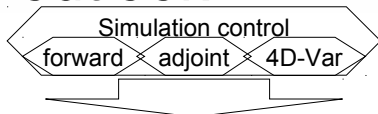
8 chemical and physical transformation modules
(6 open for operational use),
6 source terms (all open),
2 aerosol dynamics (one open)
3D- and 4D- Var
Domains: from global to beta-meso scale (~1km resolution)

Meteo input:

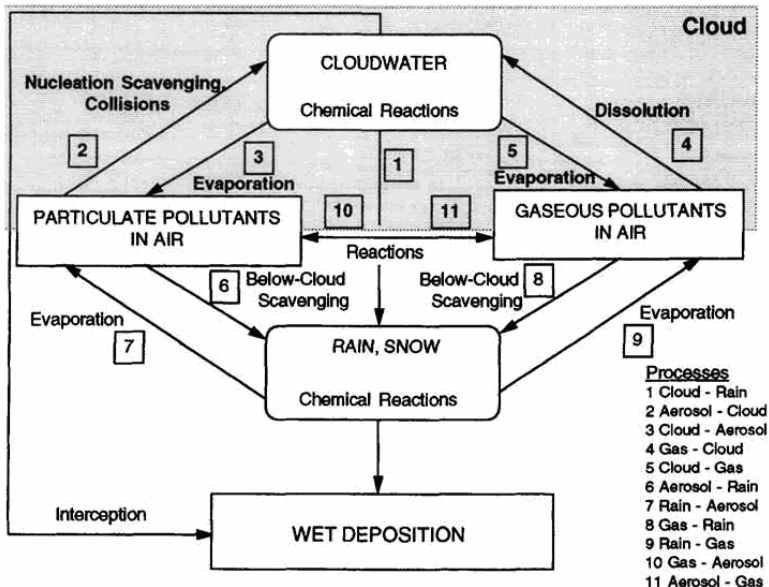
ECMWF

HIRLAM, AROME, HIRHAM,
ECHAM, and any other who can write
GRIB-1 or GRIB-2

WRF



Wet deposition



Seinfeld & Pandis (2006)

FIGURE 20.1 Conceptual framework of wet deposition processes.



Wet deposition in SILAM

Operational SILAM:

- ▶ Precipitation rate from meteo
- ▶ Prescribed cloud height
- ▶ Constant precip. below cloud
- ▶ Prescribed scavenging coefficients (in- or below-cloud, rain or snow)
- ▶ Species-dependent scaling

- ▶ Saturation of SO_2

New scheme:

- ▶ Meteo input for rain/snow profiles:
 - ▶ Precipitation rates
 - ▶ Cloud water content
 - ▶ Cloud fraction
- ▶ Equilibrium in clouds
- ▶ Fraction of cloud precipitating
- ▶ Solubilities
- ▶ Simple dissociation chemistry (SO_2)



For *most* chemicals in-water equilibrium CNC is proportional to in-air concentration

- ▶ can define “Equivalent volume” of water: in-water-mass / in-air-CNC in equilibrium
- ▶ flux \propto conc. difference
- ▶ resistance: $R = \Delta\text{CNC}/F$
- ▶ exchange is a relaxation process

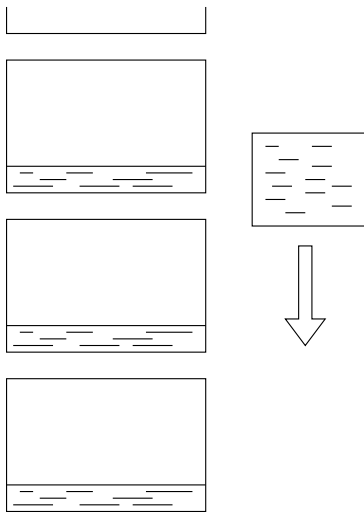
Exchange is driven by:

- ▶ equilibrium mass fractioning
- ▶ exchange time scales



Air-water relaxation

- ▶ In-cloud equilibrium
- ▶ Rainout $\tau = CWC/R$
- ▶ Washout – relaxations:
 - ▶ droplet CNC to air
 - ▶ in-air to incoming rain



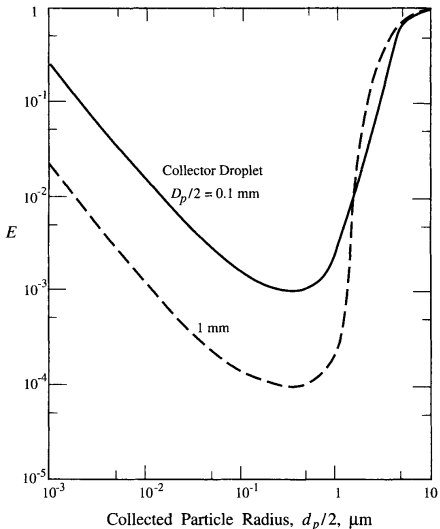
Sub-cloud scavenging efficiency



- ▶ Diffusion D
- ▶ Interception d_p
- ▶ Impaction τ_p

Accumulation around $1 \mu\text{m}$

Diffusion for gases



Seinfeld & Pandis (2006)

FIGURE 20.6 Semiempirical correlation for the collection efficiency E of two drops (Slinn 1983) as a function of the collected particle size. The collected particle is assumed to have unit density.





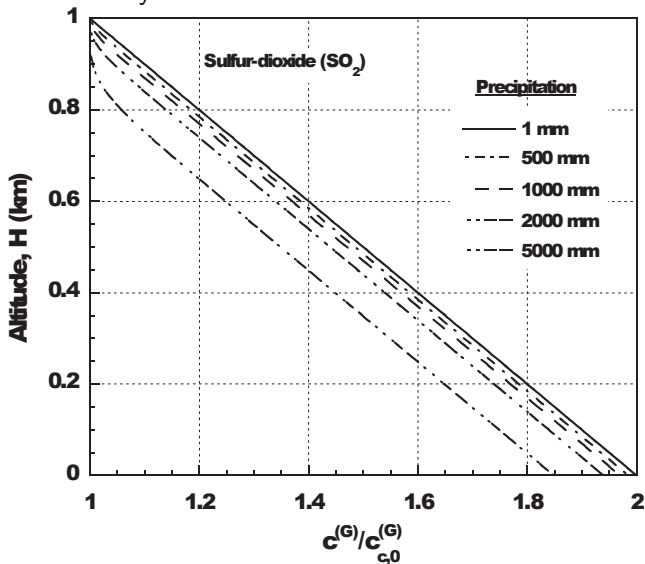
- ▶ Henry factor – mole/(l Pa) says how much of dissolved gas is in equilibrium with 1 Pa of partial pressure
- ▶ Temperature-dependent
- ▶ Henry factor does not say “how much stuff can get into droplet at given in-air concentration”
- ▶ Reason: dissociation
- ▶ Effective Henry coefficient helps helps sometimes

For SO₂ situation even worse. . .



SO₂: how NOT to

Blind use of Henry constant from manuals.



(c) Respected authors, 2011

(c) Respected journal (IF=3.7), 2011



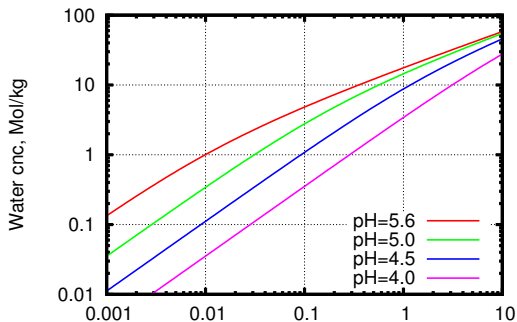
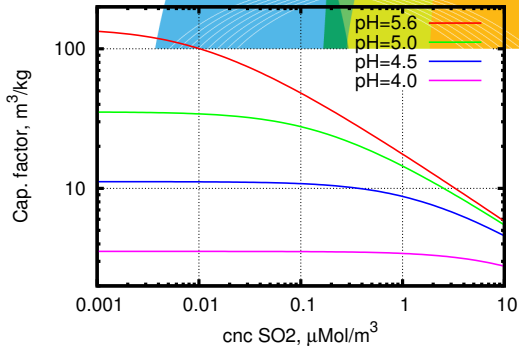
10% reduction of SO₂ in air needs 5000 mm of rain.

SO₂ solubility

Solubility depends on “background” pH of water and on SO₂ itself.

No linear models work!
(incl. scavenging coefficients)

In polluted regions
 $\sim 1 \mu\text{mol}/\text{m}^3$





The effective Henry factor for SO_2 depends on pH, in particular on the amount of dissolved SO_2

$$[\text{S(IV)}] = [\text{SO}_2] \left(1 + \frac{K_{S1}}{[\text{H}^+]} \right)$$

Result: Solubility concept does not apply.

Operational SILAM: Saturation of SO_2 in rain water.

New scheme – approximate electro-neutrality:

$$[\text{H}^+] \simeq [\text{Strong Acids}] + [\text{HSO}_3^-] - [\text{NH}_4^+], \quad [\text{HSO}_3^-] \simeq [\text{S(IV)}].$$



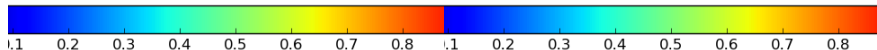
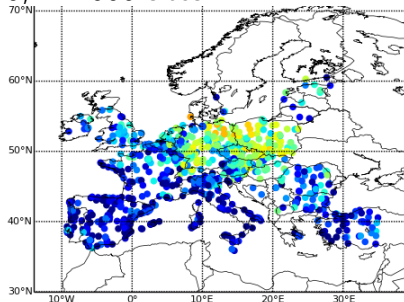
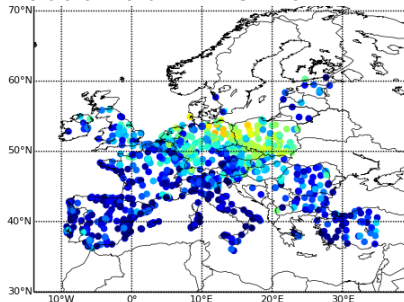


- ▶ SO₂ last species to scavenge
- ▶ In-water acids and ammonia are known
- ▶ Effective equilibrium solubility for given amount of water and SO₂ in cell, and given acids in rain.
- ▶ Same linear algorithm is asymptotically right



Improvement

Correlation oldWD vs. newWD. 2010, ~ 1000 sites



	OLD	RMS	Bias	Corr.	NEW	RMS	Bias	Corr.
95th		11.30	1.46	0.58		11.16	1.82	0.64
Mean		3.94	-1.23	0.26		3.84	-0.90	0.31
Median		2.97	-0.87	0.27		2.94	-0.59	0.32





Example:

- ▶ heavy rain in half of grid cell
- ▶ rain scavenges 99% per time step
- ▶ little wind

Question:

- ▶ How much stuff will be left in air after two time steps?





- ▶ WD scheme based on relaxation
- ▶ Incloud and subcloud
- ▶ Dissociation must be accounted for SO₂
- ▶ Substantial improvement of SO₂ concentrations
- ▶ Subgrid effects need to be accounted for

