

CFD Modelling of Reactive Pollutants Dispersion and Effect of Photocatalytic Pavements in a Real Urban Area

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Outline

1. Introduction

2. Experimental Campaign

3. CFD Model Description and Set-Up

4. Results

- Evaluation of the chemical effects on pollutants dispersion with experimental data
- Analysis of the Photocatalytic Effect

5. Conclusions

1. Introduction

- Exhaust gases are the main source of NO and NO₂ emissions in an urban area



Worsen of Urban Air Quality



- An accurate understanding of urban air quality requires considering the coupled behavior between dispersion of reactive pollutants and atmospheric dynamics.
- Usually, NO and NO₂ are modeled as passive tracer at microscale.

Which is the impact on NO and NO₂ concentrations by including chemical reactions in a CFD model in a real urban zone?

- The behavior of the photocatalytic materials has been studied extensively in controlled laboratory conditions and they are being considered as a possible solution to reduce NO_x concentrations in urban areas.

Which is the efficiency of this material in real urban areas?

- Within the framework of LIFE MINOX-STREET Project, the efficiency of photocatalytic materials is being researched in real urban scenarios.

Main Objective: CFD Modelling of NO and NO₂ dispersion applying different chemical approaches including the NO deposition effect by photocatalytic pavement in a real urban area

1. Evaluation of the chemical effects on NO and NO₂ dispersion



Comparison with experimental measurements

2. Analysis of the photocatalytic effect on NO concentration

2. Experimental Campaign

- Location: North of Madrid
- 25th September - 25th October → The maximum effectiveness of NO deposition by Photocatalytic Materials is obtained under specific meteorological conditions

$$\left. \begin{array}{l} R > 400 \text{ Wm}^{-2} \\ RH < 65 \% \\ U < 5 \text{ m s}^{-1} \end{array} \right\}$$

29th September, 2015 → 12.00-13.00 UTC



- Background Measurements (●) → $\left\{ \begin{array}{l} h = 20 \text{ m} \\ d = 300 \text{ m} \end{array} \right.$
- Wind speed and direction
- Pollutants concentration: NO, NO₂ and O₃

In the research area:

- Photocatalytic area

Laboratory Tests

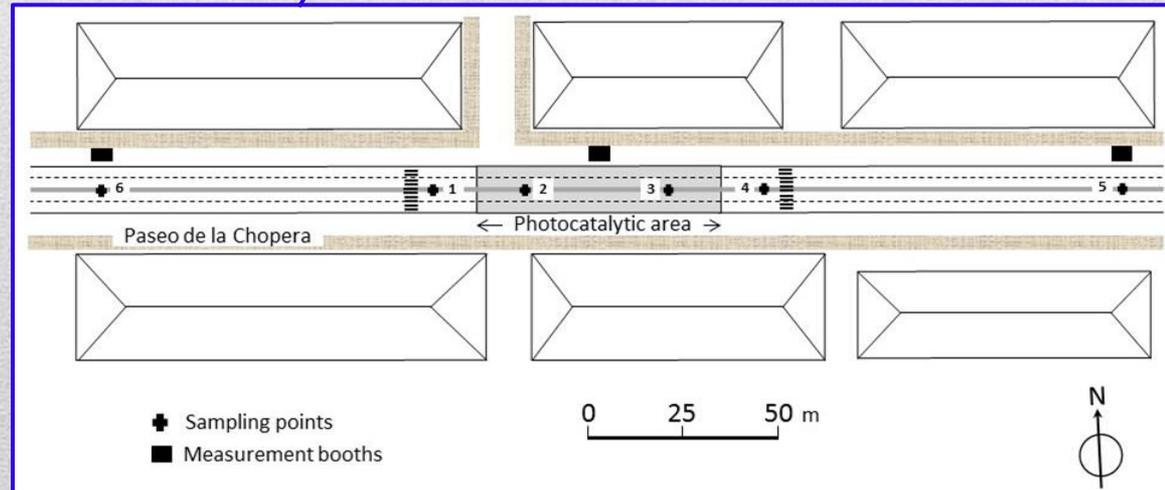
- $L = 60 \text{ m}$

- NO deposition: $V_d = 0.5 \text{ cm s}^{-1}$

- Measurements Points

- 6 sampling points: NO and NO₂

- $h = 1 \text{ m}$

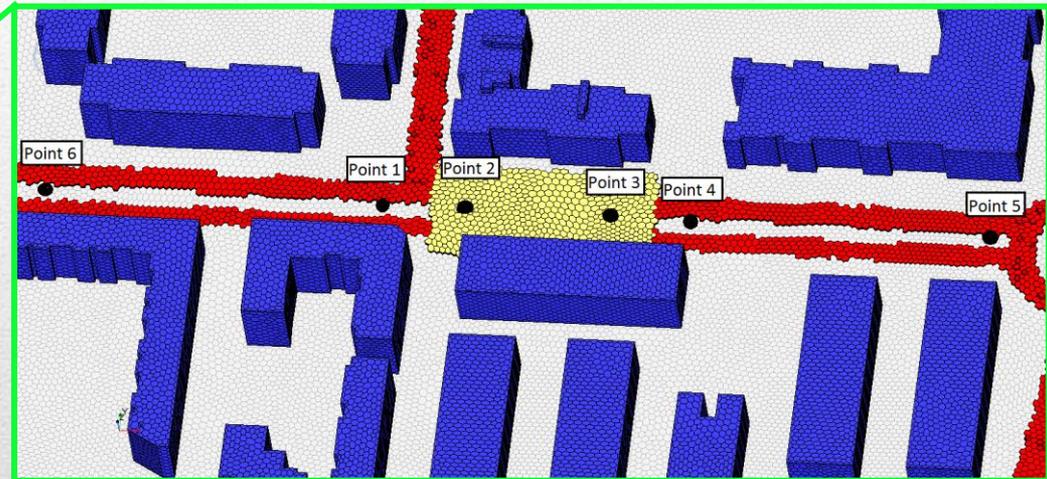
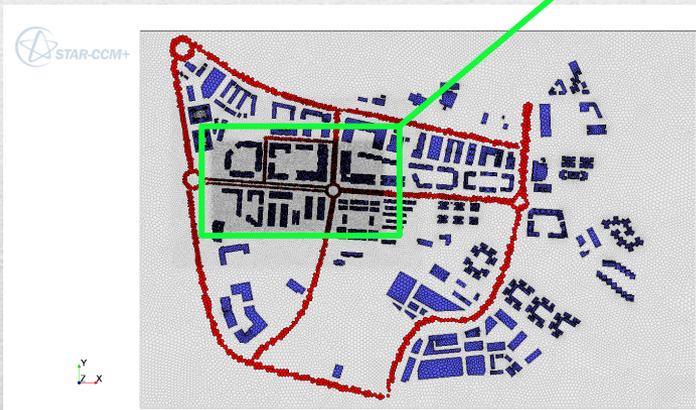


More details in the poster session of this conference (Pujadas et al. (ID. 090))

3. CFD Model Description and Simulations Set-Up

Numerical simulations are based on the Reynolds averaged Navier-Stokes equations (RANS) with the k- ϵ turbulence model (STARCCM+ v9.04.011-R8)

- Domain: 1.5 km x 1 km
 - Building height ~ 20m



- Polyhedral Mesh

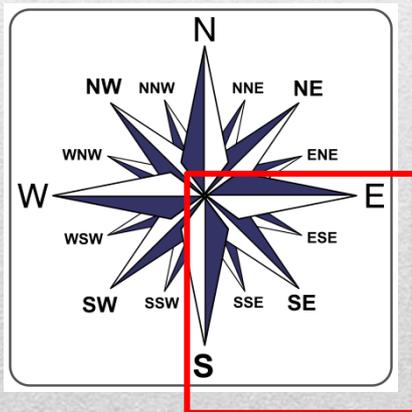
Out ~ 6 m

In ~ 2m with refine regions (<1 m)



$2.3 \cdot 10^6$ grid cells

- Unsteady state simulations
- Inlet boundary conditions from experimental data (●)
 - At roof of the building (h=20 m)
 - $d=300$ m
 - $\Delta t = 5min$
- Meteorological conditions → Neutral atmospheric conditions



$$u_{in}(z) = \frac{u_*}{\kappa} \ln\left(\frac{z}{z_0}\right)$$

$$k_{in} = \frac{u_*^2}{C_\mu^{1/2}}$$

$$\varepsilon_{in}(z) = \frac{C_\mu^{3/4} k_{in}^{3/2}}{\kappa z}$$

u_* : Friction Velocity
 C_μ : Constant (0.9)
 z_0 : Roughness length
 κ : von Karman constant (0.4)

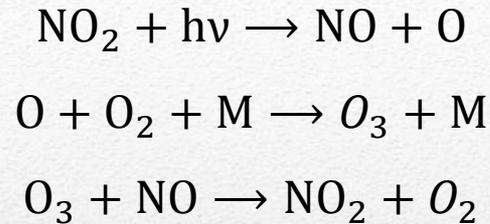
- Background concentration: NO, NO₂ and O₃ →

	Values range
NO	[5.06 – 6.98] ppb
NO ₂	[9.16 – 11.99] ppb
O ₃	[39.60 – 46.63] ppb

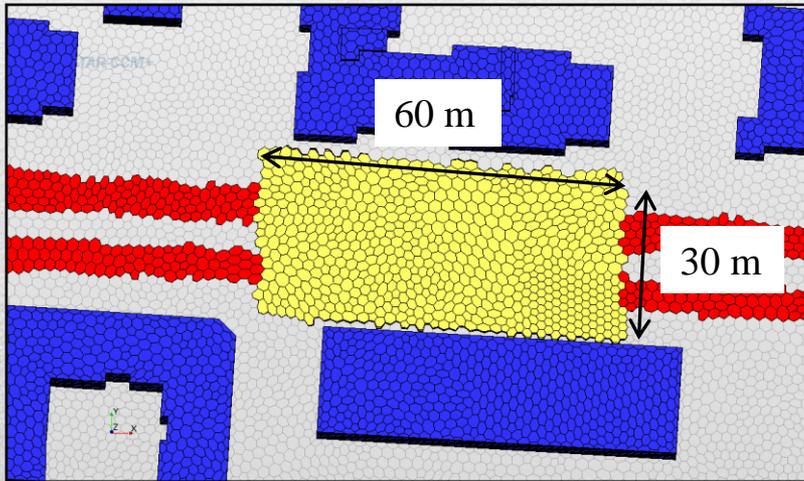
Chemical approaches

- Non-Reactive pollutants

- Photostationary Steady State (PSS):



Photocatalytic effect

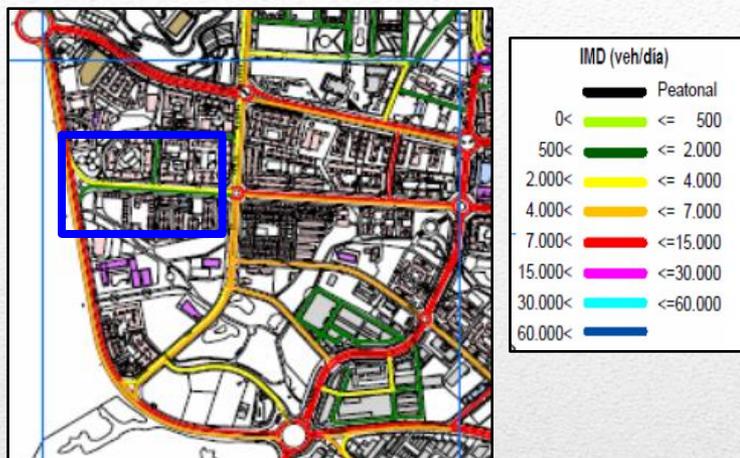


- Sink of NO: $dep_{\text{NO}} = -[\text{NO}] \cdot V_d$
- $V_{d,exp} = 0.5 \text{ cm s}^{-1} \leftarrow \text{Laboratory Tests}$

NO Transport Equation

$$\frac{\partial \text{NO}}{\partial t} + U_i \frac{\partial \text{NO}}{\partial x_j} = D \frac{\partial^2 \text{NO}}{\partial x_j \partial x_j} + \frac{\partial}{\partial x_j} \left(K_c \frac{\partial C_i}{\partial x_j} \right) + [\Delta \text{NO}]_{chem} + S_{em} + dep_{\text{NO}}$$

■ NO_x Emission



○ Emission Factor (EF):

Vehicle type	NO _x (g/km)
Bus	3,46
Motorbike	0,13
Vehicle	0,44
Light vehicle	0,81
Heavy vehicle	1,86

- $NO + NO_2 = NO_x$
- Volumetric emission ratio (*): $\frac{NO}{NO_2} = 10$

○ Within the studied area:

- No. of vehicles
- Vehicle type

The emission changes every 5 min

$$\rightarrow S_{NOx} = EF_{type_veh} \cdot N_{veh}$$

○ Outside the studied area:

$$S_{NOx}(out) = TN_{veh,RS} \frac{DTI_{out}}{DTI_{RS}} EF_{veh}$$

$TN_{veh,RA}$: Total number of vehicles in the research street

DTI_{RS} : DTI in the research street

(*) Baker et al., 2004

4. Results

1. Evaluation of the simulated chemical approaches



Differences in the simulated concentration of NO and NO₂ regarded as:

- Non-reactive pollutants
- Photostationary Steady State



1.1. Spatial distribution concentration

1.2. Time series at measurements points



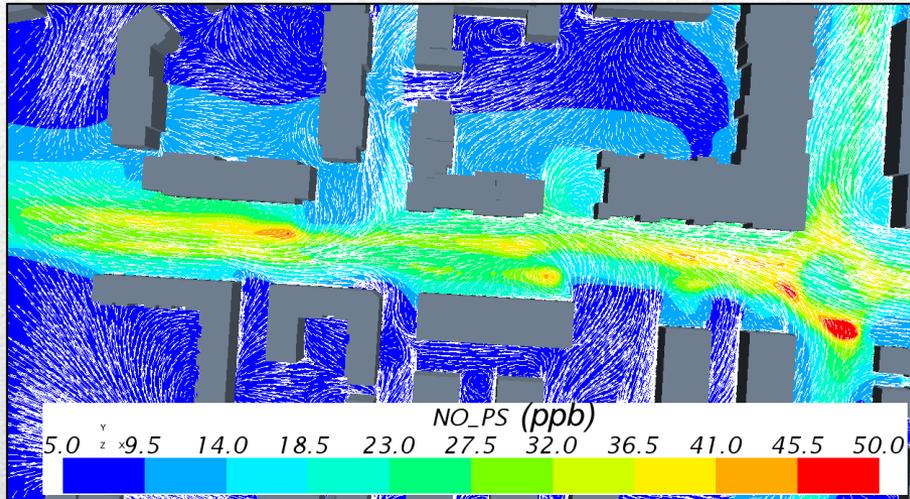
1.3. Evaluation of the time average concentration using experimental data

2. Study of the photocatalytic effect using simulation results in a real urban scenario

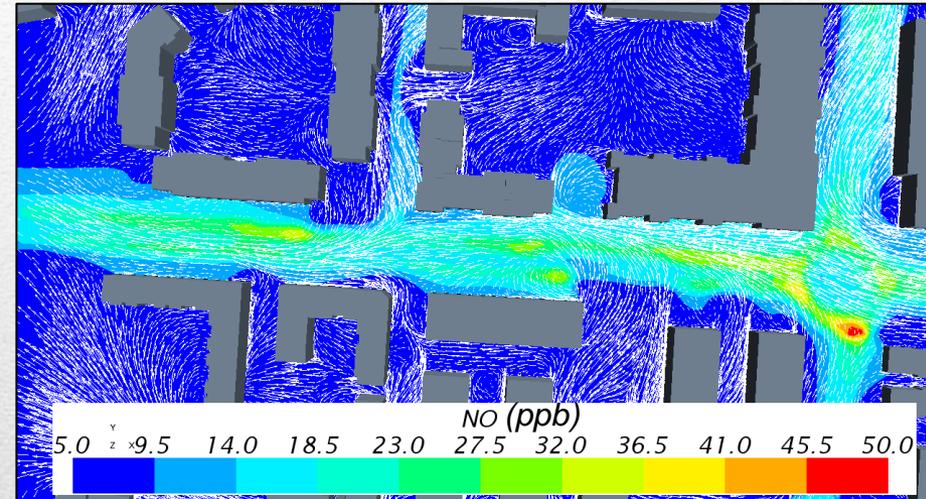
4.1. Evaluation of the simulated chemical approaches

- Spatial distribution at h=1 m → t=60 min

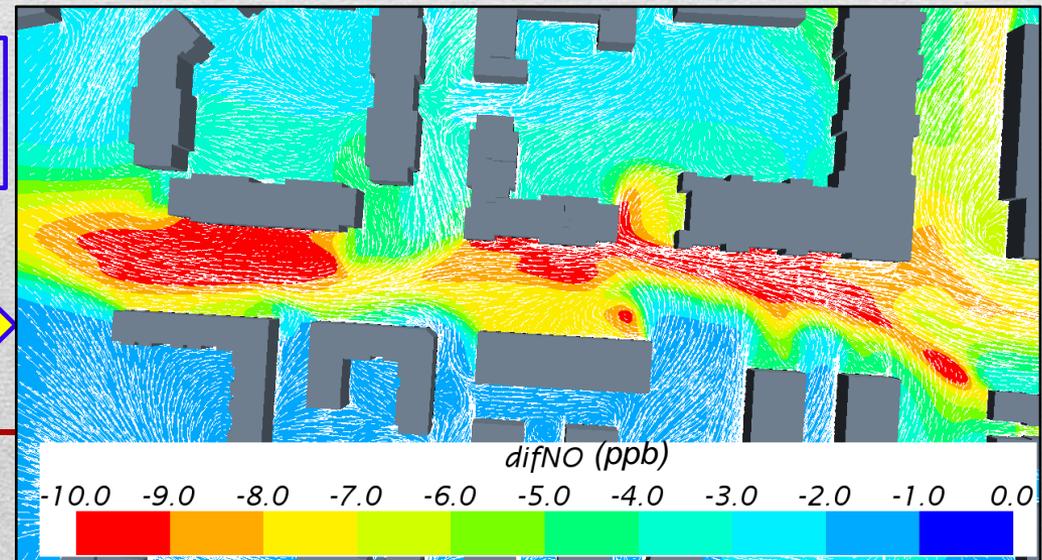
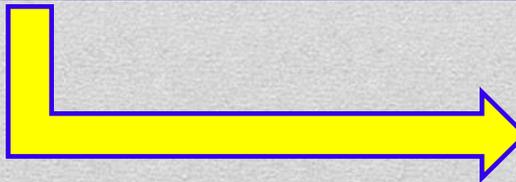
NO tracer



NO reactive

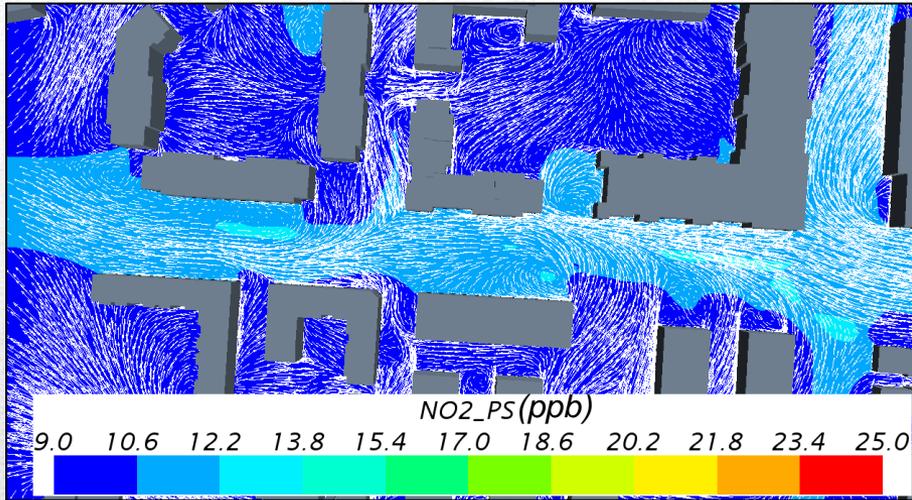


$$difNO = NO_{reactive} - NO_{tracer}$$

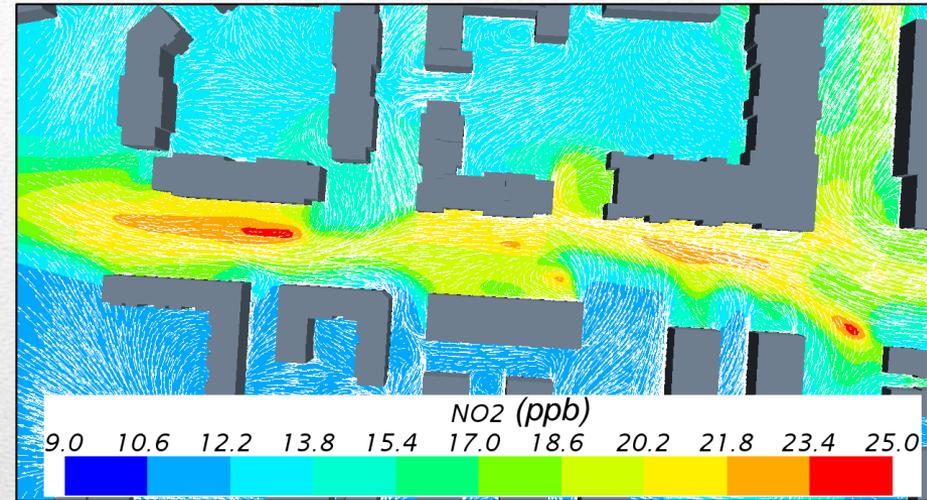


- Spatial distribution at h=1m → t = 60 min

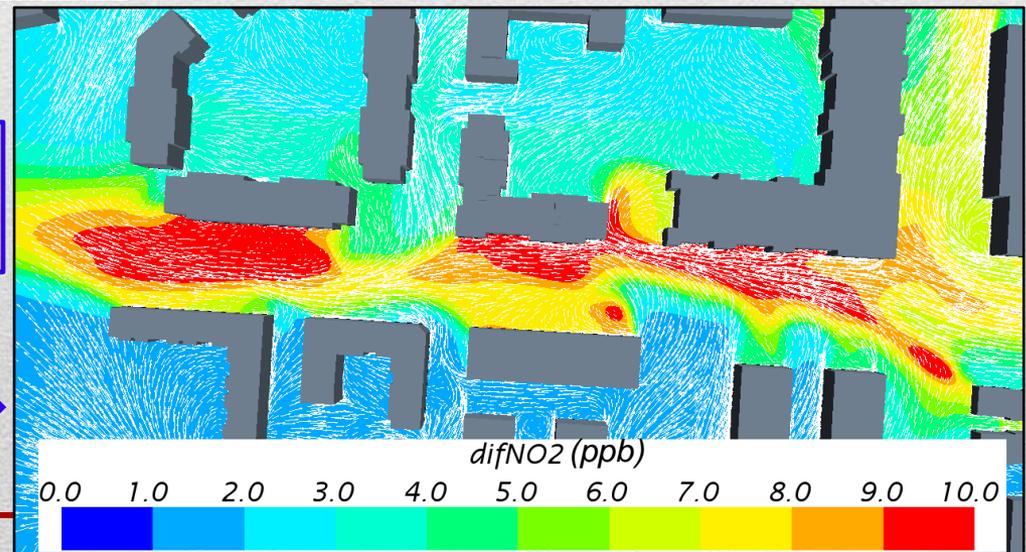
NO₂ tracer



NO₂ reactive

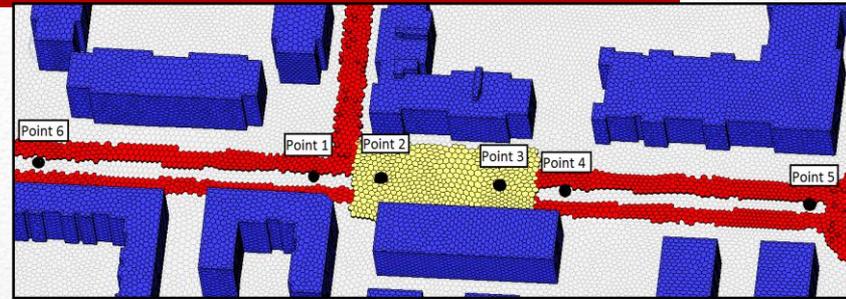


$$difNO_2 = NO_{2, reactive} - NO_{2, tracer}$$

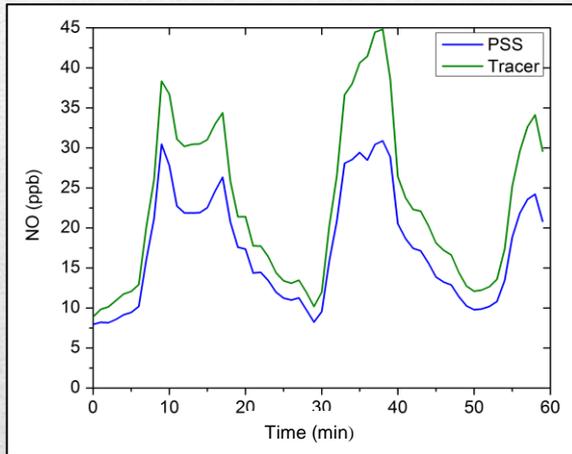


Time series of simulated NO

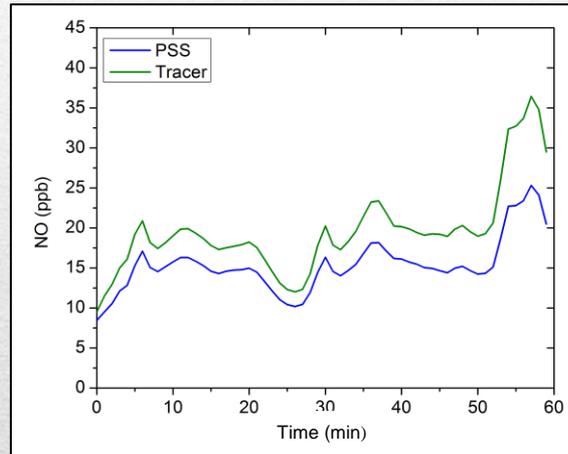
Differences Tracer and Reactive \rightarrow Location
 \rightarrow Atmospheric conditions



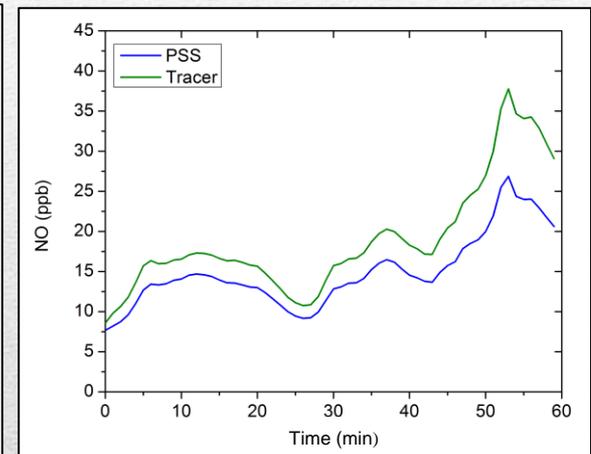
Point 1



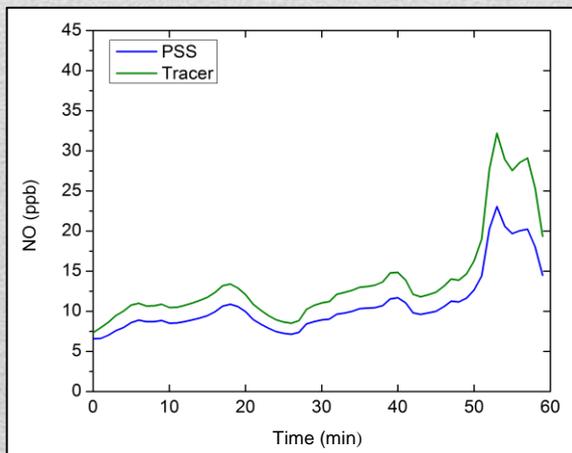
Point 2



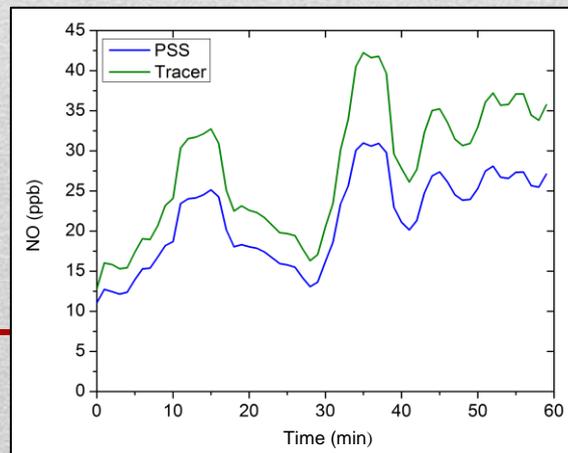
Point 3



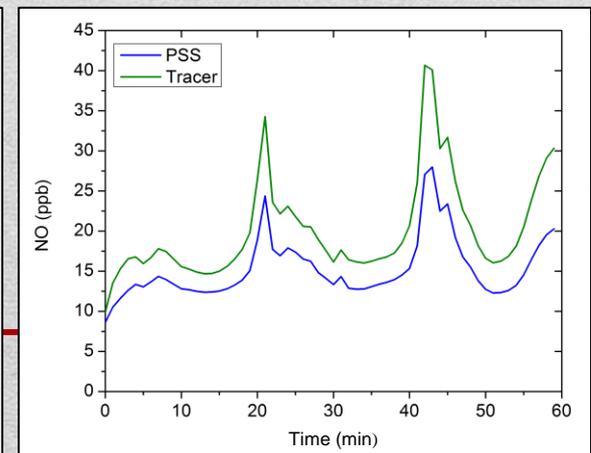
Point 4



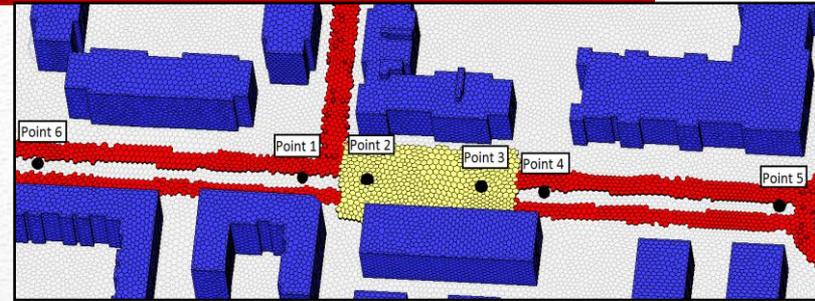
Point 5



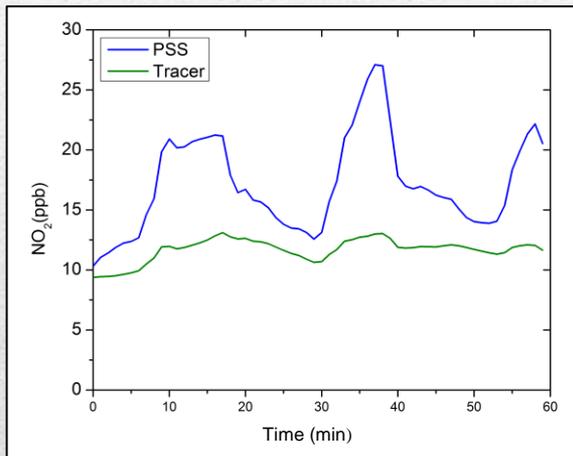
Point 6



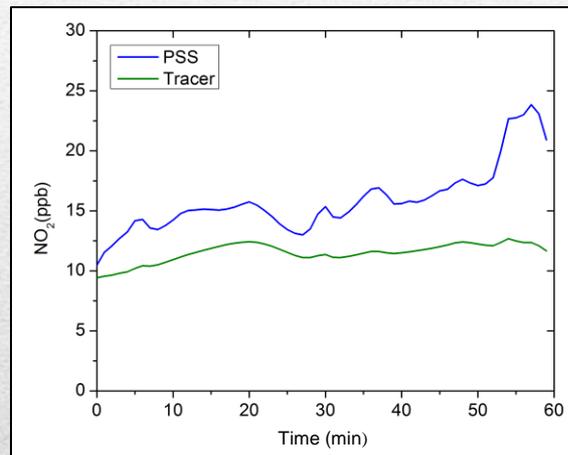
■ Time evolution of NO₂



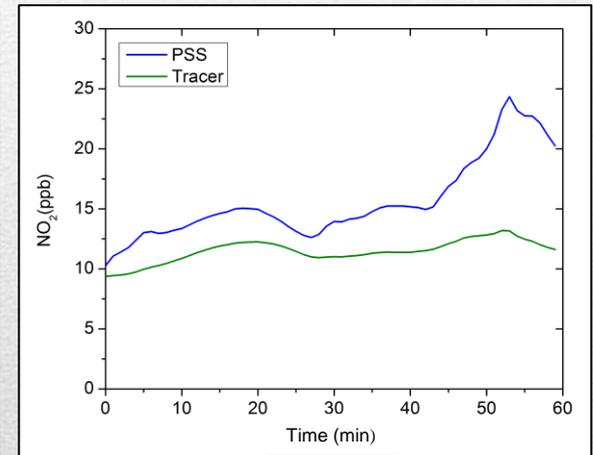
Point 1



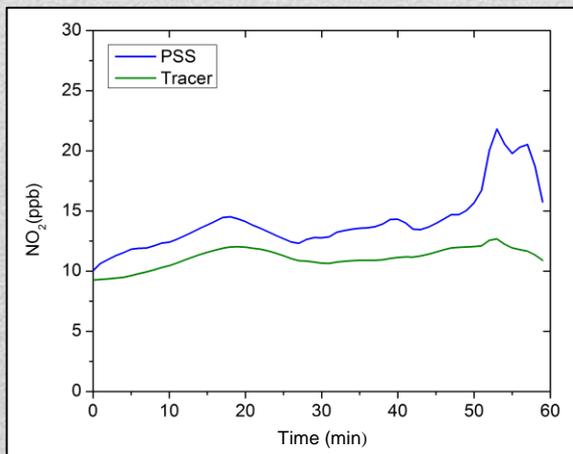
Point 2



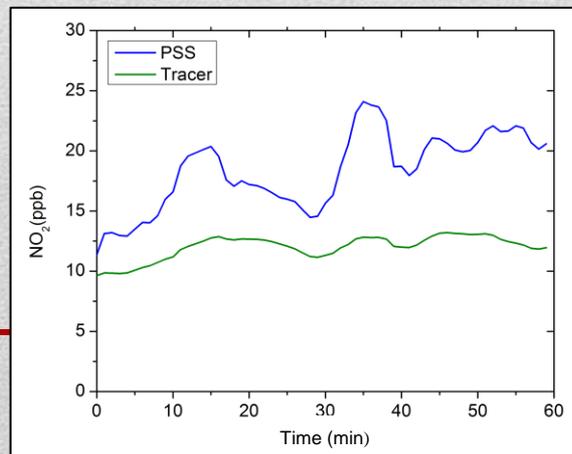
Point 3



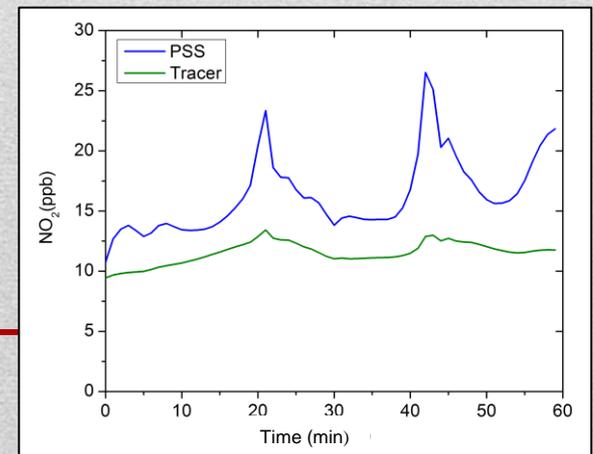
Point 4



Point 5

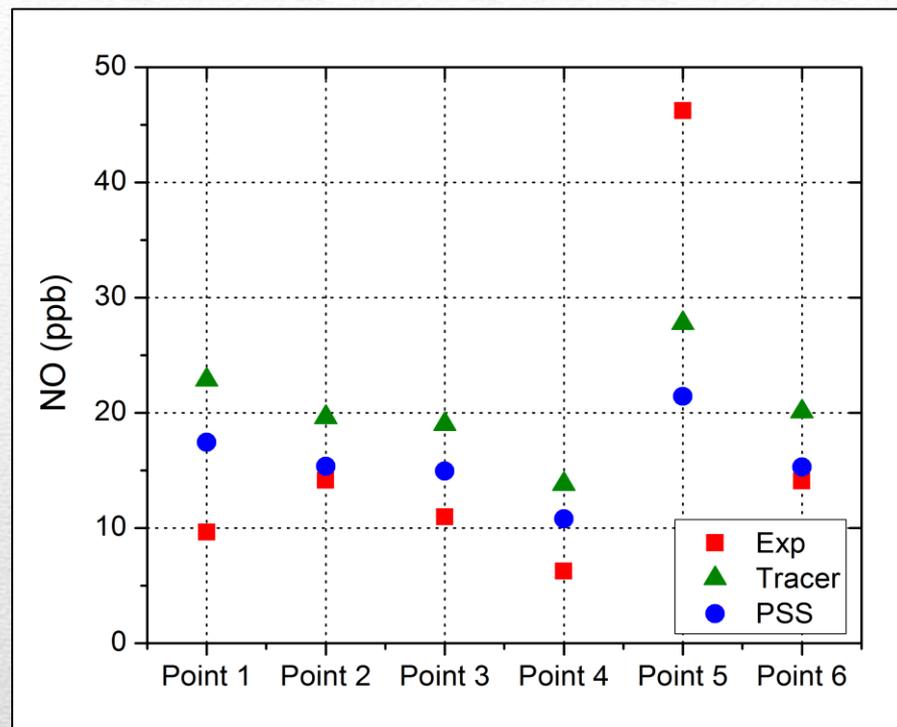


Point 6



- Comparison of the time average concentration with experimental measurements

- NO



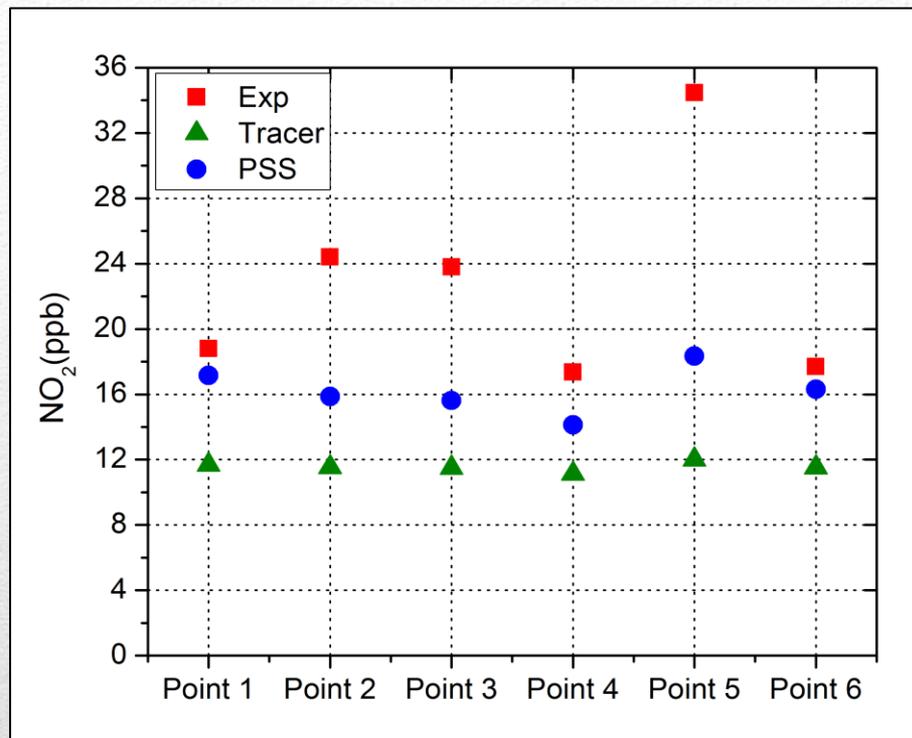
○ Including a chemical scheme



Better fit of NO concentration

	Tracer	PSS	Acceptance Criteria (Goricsan et al., 2011 and Chang et al., 2005)
NMSE	0.30	0.39	NMSE < 1.5
FB	-0.20	0.06	-0.3 < FB < 0.3
FAC2	66.6%	83.3%	

■ NO₂



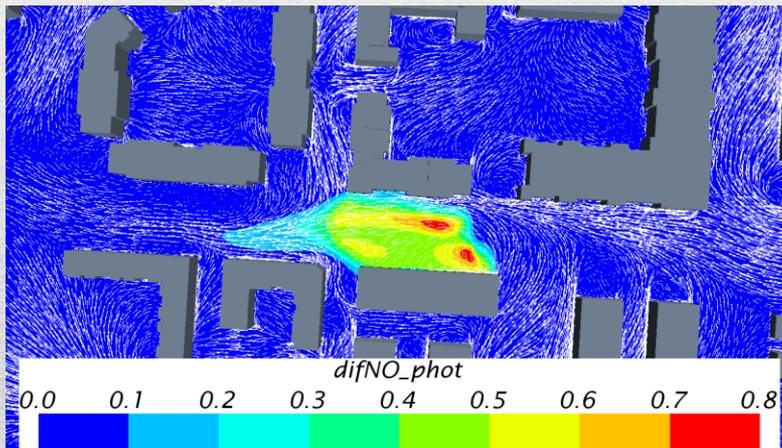
- Underestimation of the NO₂ concentration
- Better results using the photostationary steady state

	Tracer	PSS	Acceptance Criteria (Goricsan et al., 2011 and Chang et al., 2005)
NMSE	0.55	0.18	NMSE < 1.5
FB	0.65	0.33	-0.3 < FB < 0.3
FAC2	50 %	100 %	

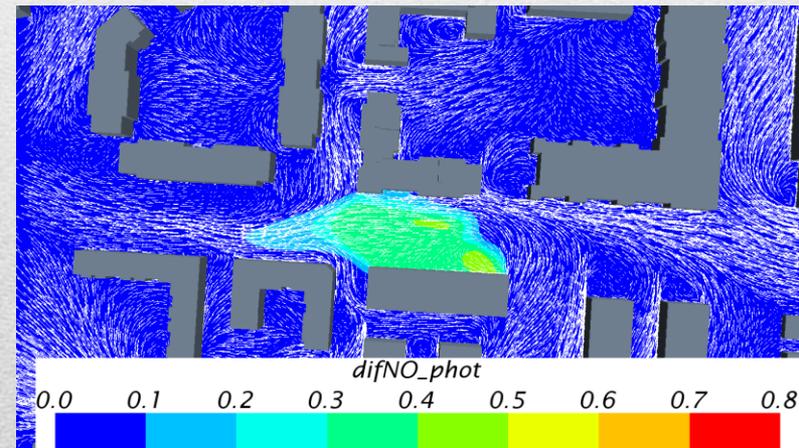
4.2. Analysis of the photocatalytic effect in the real urban scenario

- Photostationary steady state is selected
- 2 CFD simulations with same conditions
 - With Photocatalytic pavement
 - Without Photocatalytic pavement
- Spatial distribution at $t=60$ min → $difNO = NO - NO_{dep}$

$h=1m$



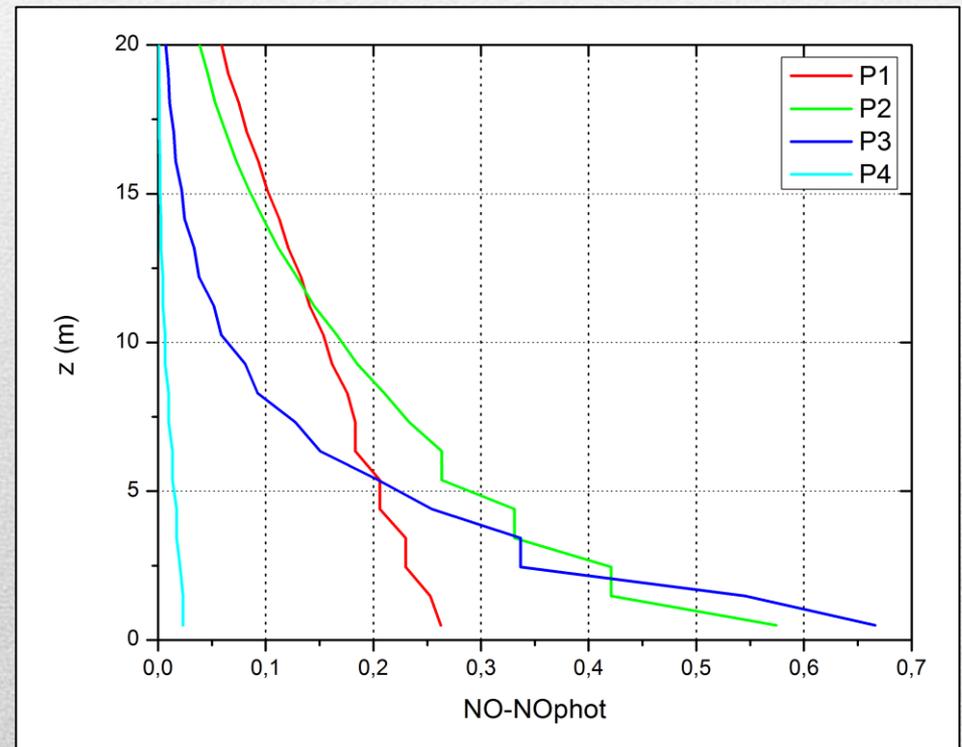
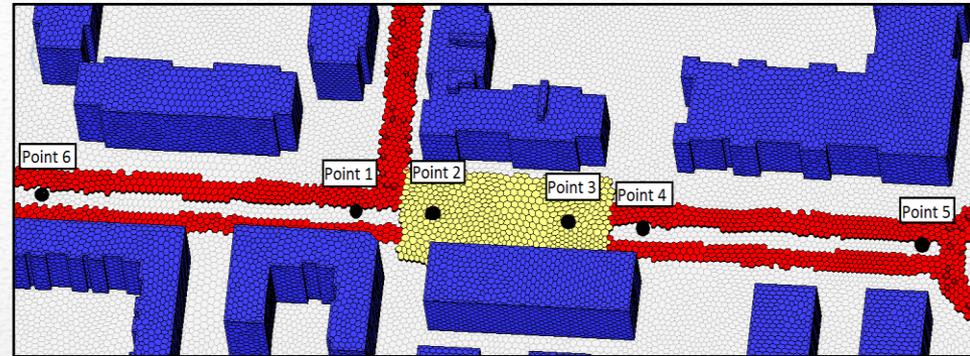
$h=3m$



- The photocatalytic effect is negligible and it is only observed over the pavement.
- The reduction in NO concentration at 1 m is slightly higher than at pedestrian level

Vertical Profiles of the decrease of NO concentration due to photocatalytic pavement

- The maximum value is 0.7 ppb.
- The maximum differences are found in the points located over the pavement



Conclusions

- The introduction of chemical reactions in the CFD simulations modifies the amount of pollutant concentration so that the NO concentration is reduced and NO₂ concentration is increased.
- NO and NO₂ concentration simulated by both chemical approaches are in agreement with the experimental data.
- Better results of NO and NO₂ concentration are obtained taking into account reactive pollutants using the photostationary steady state.
- The photocatalytic effect is evaluated by means of CFD simulations considering reactive pollutants and the NO deposition due to photocatalytic pavement. The results show a small decrease in NO concentration, even close to the material at ground.
- These results are obtained for a selected photocatalytic material in specific meteorological conditions in a real urban area.



Thank you



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