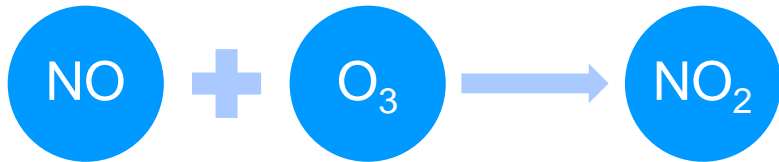
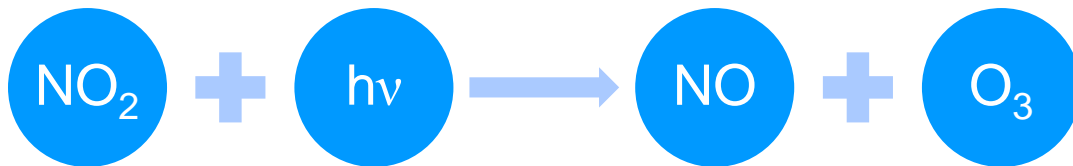


# A validation study of the ADMS plume chemistry schemes



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Atmospheric Dispersion Modelling for Regulatory Purposes  
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This study was funded by **BP International Limited.**

# Where does NO<sub>2</sub> come from?

- Downwind of an industrial source, NO<sub>2</sub> arises from three main contributions:

### Long-range transport & long timescale chemistry

- Ambient background level of NO<sub>2</sub> in the atmosphere.
- Long-range transport of emissions and long timescale chemical reactions.

### Primary emissions of NO<sub>2</sub>

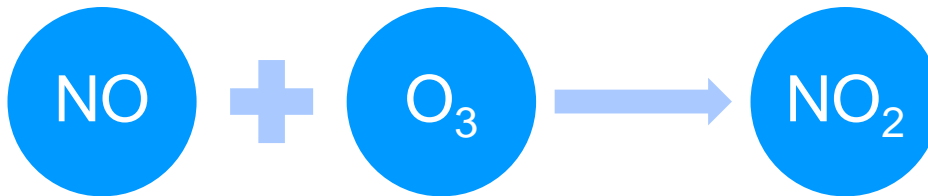
- A proportion of NO<sub>x</sub> is emitted as NO<sub>2</sub>.
- Proportion depends on source type, typically between 5% and 25% for industrial sources.

### Secondary NO<sub>2</sub> from short timescale chemical reactions

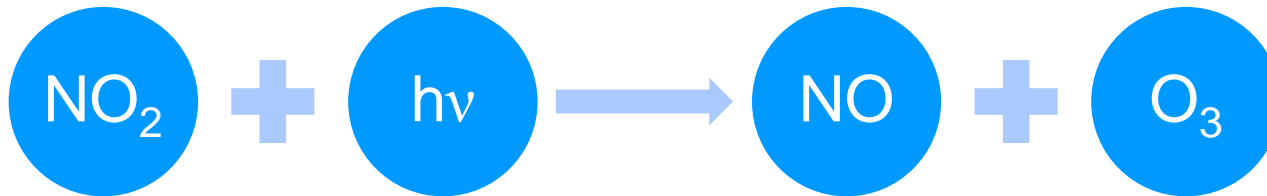
- Away from local sources of NO<sub>x</sub>, the background proportion of NO<sub>2</sub> is typically 70% - 95%.
- Secondary NO<sub>2</sub> comes primarily from the oxidation of NO.

## NO<sub>x</sub> chemistry reactions

- Two short timescale chemical reactions are considered
  - The creation of NO<sub>2</sub>:



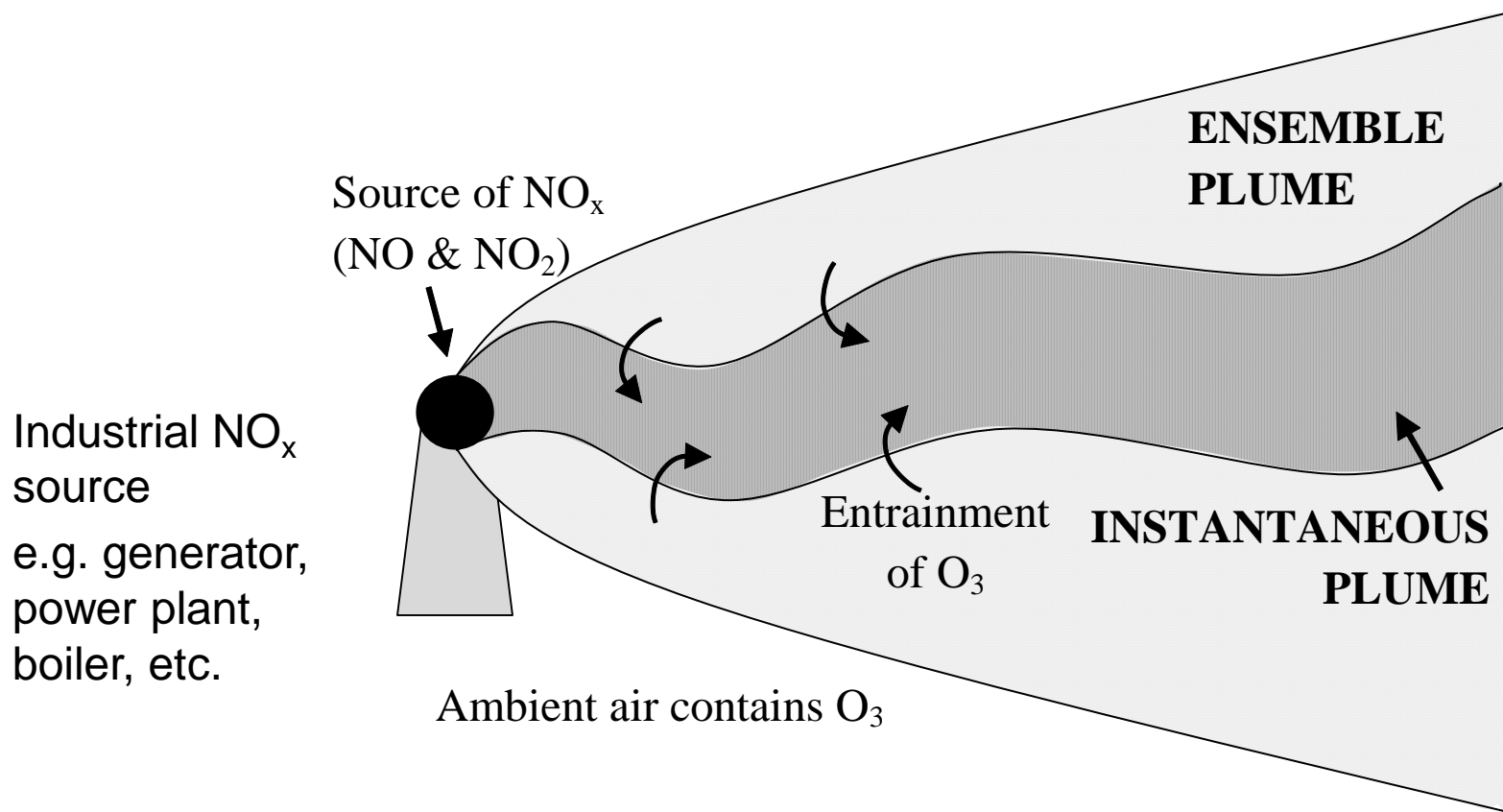
- The photolysis of NO<sub>2</sub> by sunlight:



- O<sub>3</sub> comes from atmospheric background – not emitted from sources
- Other chemical reactions occur, but over longer timescales and with species less abundant in an industrial setting.

# Modelling NO<sub>x</sub> chemistry

## Modelling a plume



# NO<sub>x</sub> chemistry modelling schemes

## Ensemble plume chemistry

- Assumes O<sub>3</sub> well mixed along whole plume
- Chemistry occurs in ensemble plume
- Conservative prediction of NO<sub>2</sub>

## Instantaneous plume chemistry

- Accounts for amount of background O<sub>3</sub>, NO<sub>x</sub> and NO<sub>2</sub> entrained into plume
- Chemistry occurs in instantaneous plume
- More theoretically accurate prediction of NO<sub>2</sub>

# Considerations when validating NO<sub>x</sub> chemistry

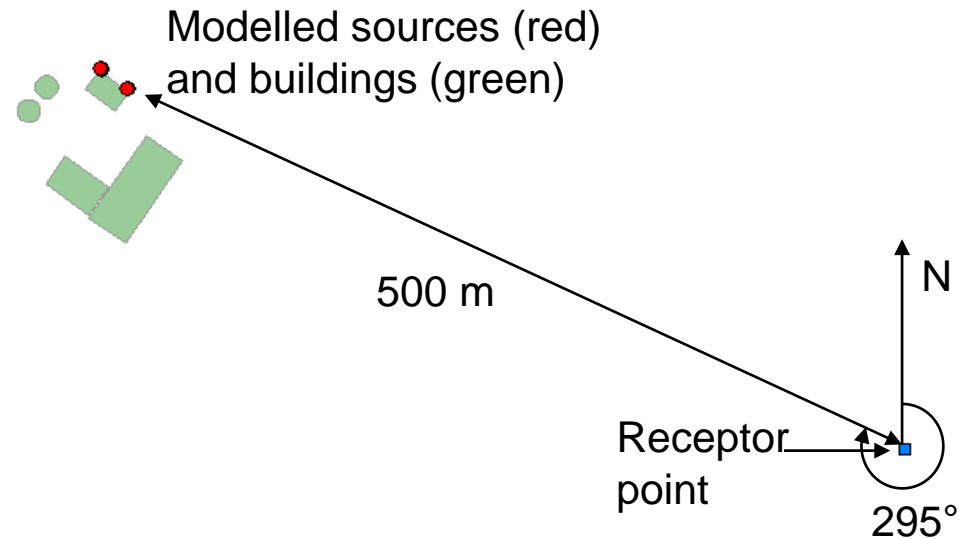
- Two aspects should be considered for NO<sub>2</sub> validation:
  - Dispersion processes – accuracy of NO<sub>x</sub>
  - Chemistry processes – accuracy of NO<sub>2</sub>
- Not possible to determine NO<sub>2</sub> accuracy independently of NO<sub>x</sub> accuracy
- NO<sub>2</sub> performance must be considered in relation to NO<sub>x</sub> validation
  - Ideally NO<sub>2</sub> performance would be similar to NO<sub>x</sub> performance
  - Cannot expect NO<sub>2</sub> performance to be good if NO<sub>x</sub> performance is poor

## Note

- Restrict comparisons to appropriate wind directions and hours, when source plume impacts monitors

## Case study: Wainwright

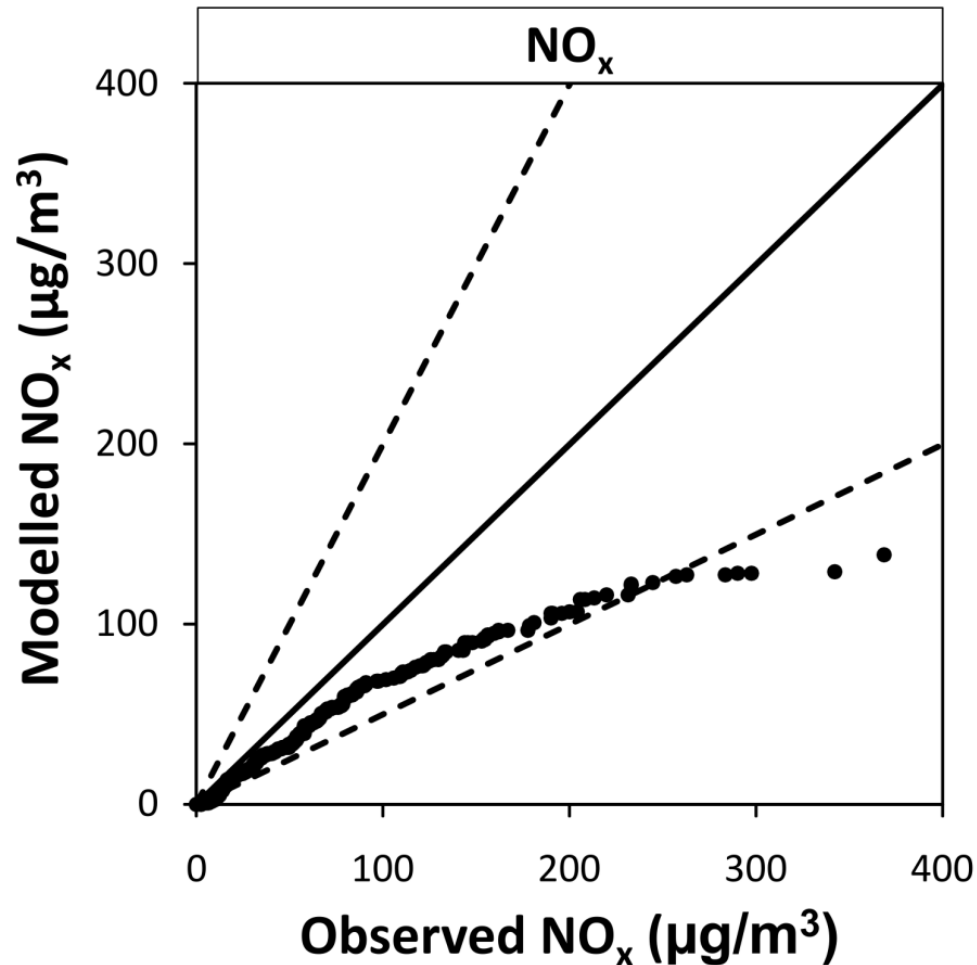
- Validation of ADMS 5 NO<sub>x</sub> chemistry with Wainwright dataset:
  - Small, diesel powered power plant in Alaska
  - 5 short stacks, similar to building height, mostly single stack operation
  - Measurement campaign over 12 ½ months
  - Single receptor point





# Case study: Wainwright

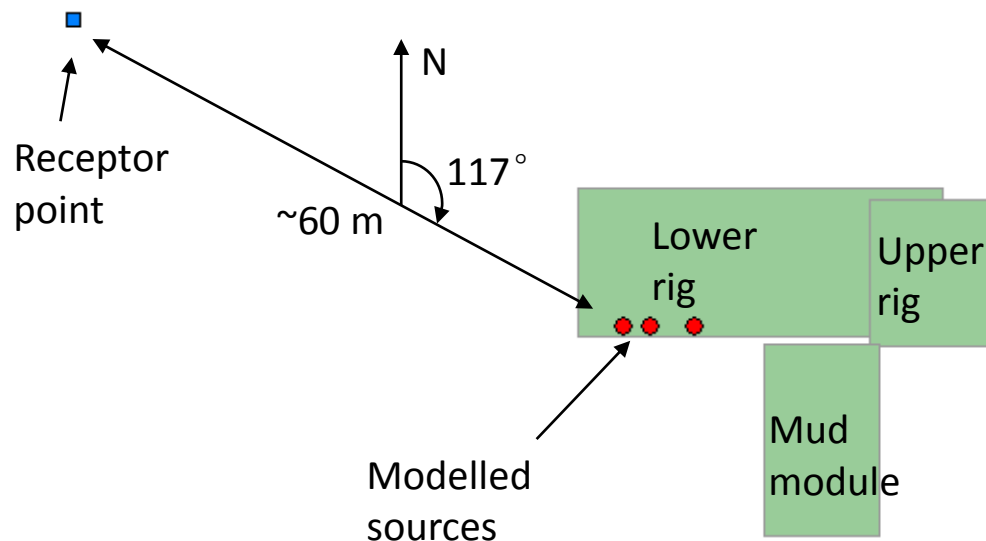
NO<sub>x</sub> results (µg/m<sup>3</sup>)



Observed Mean	43
Modelled Mean	26
R	0.79
Fac2	0.52
Observed Max	370
Modelled Max	140

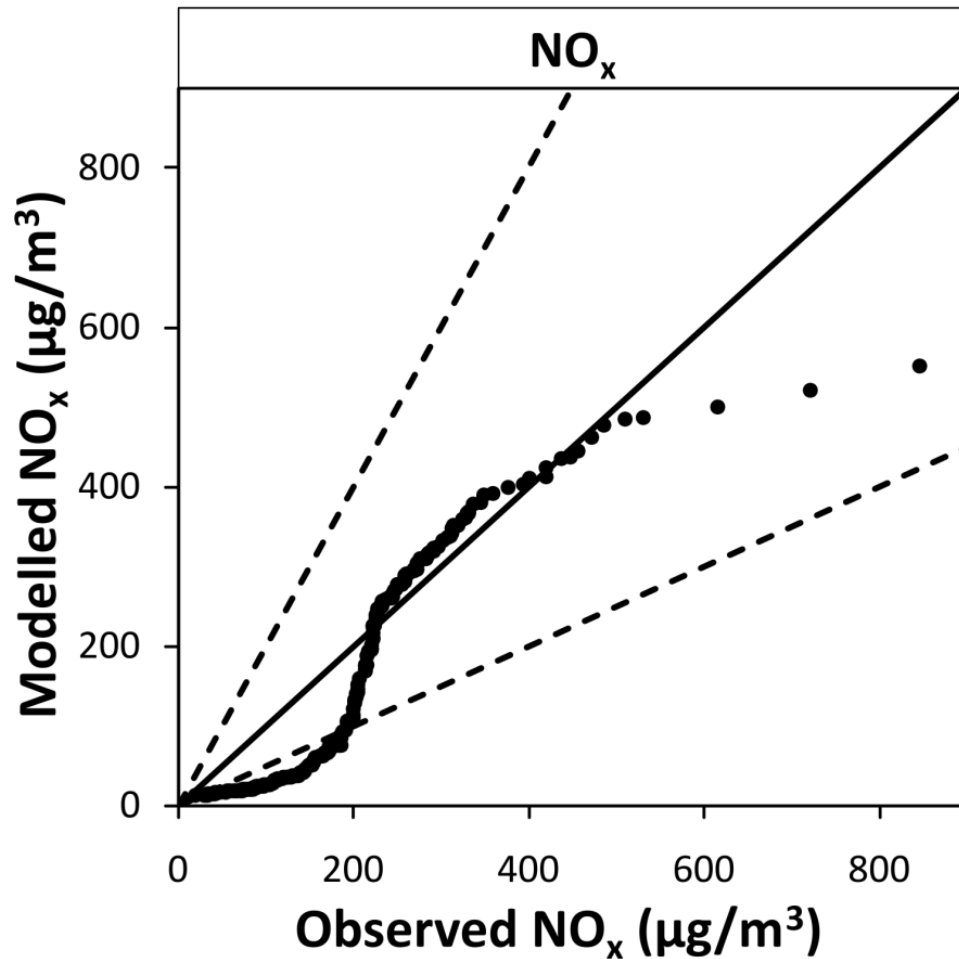
## Case study: Prudhoe Bay

- Oil drilling rig on the North Slope of Alaska
- Measurement campaign over 40 days
- Three largest sources modelled
- One monitor, very close to sources
- Measured  $\text{NO}_x$ ,  $\text{NO}_2$  and  $\text{O}_3$  concentrations
- Measured met conditions



# Case study: Prudhoe Bay

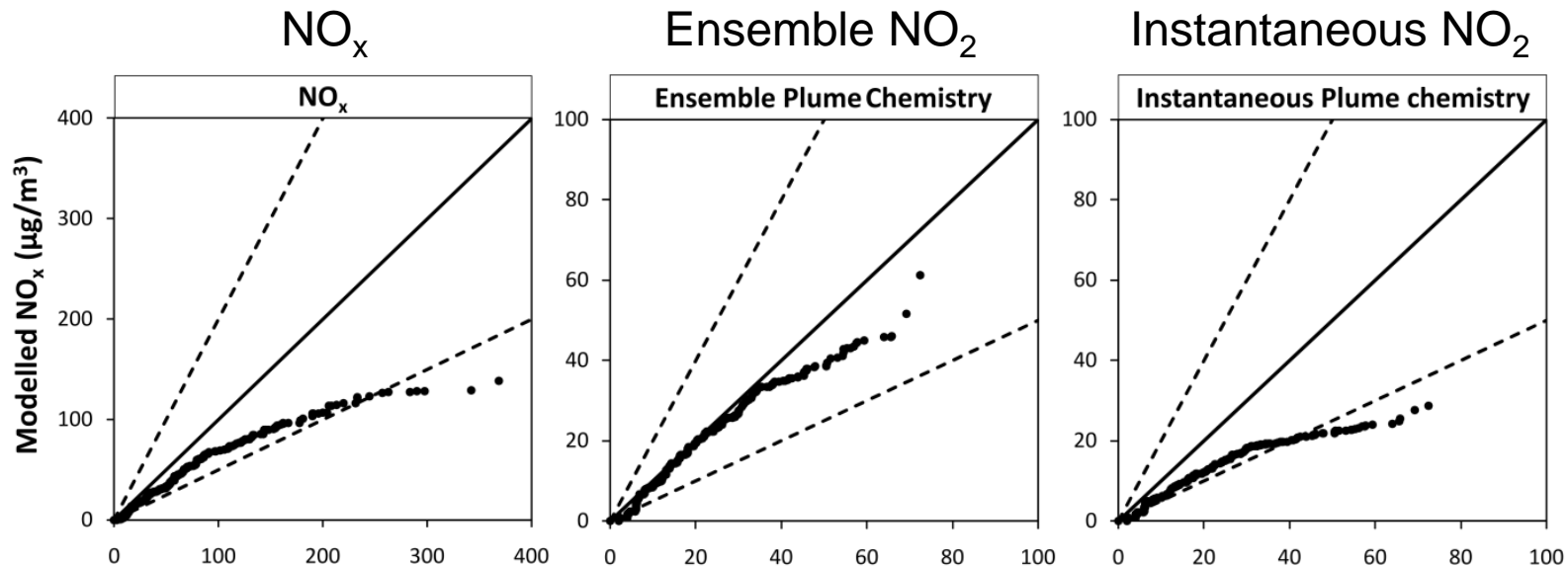
NO<sub>x</sub> results (µg/m<sup>3</sup>)



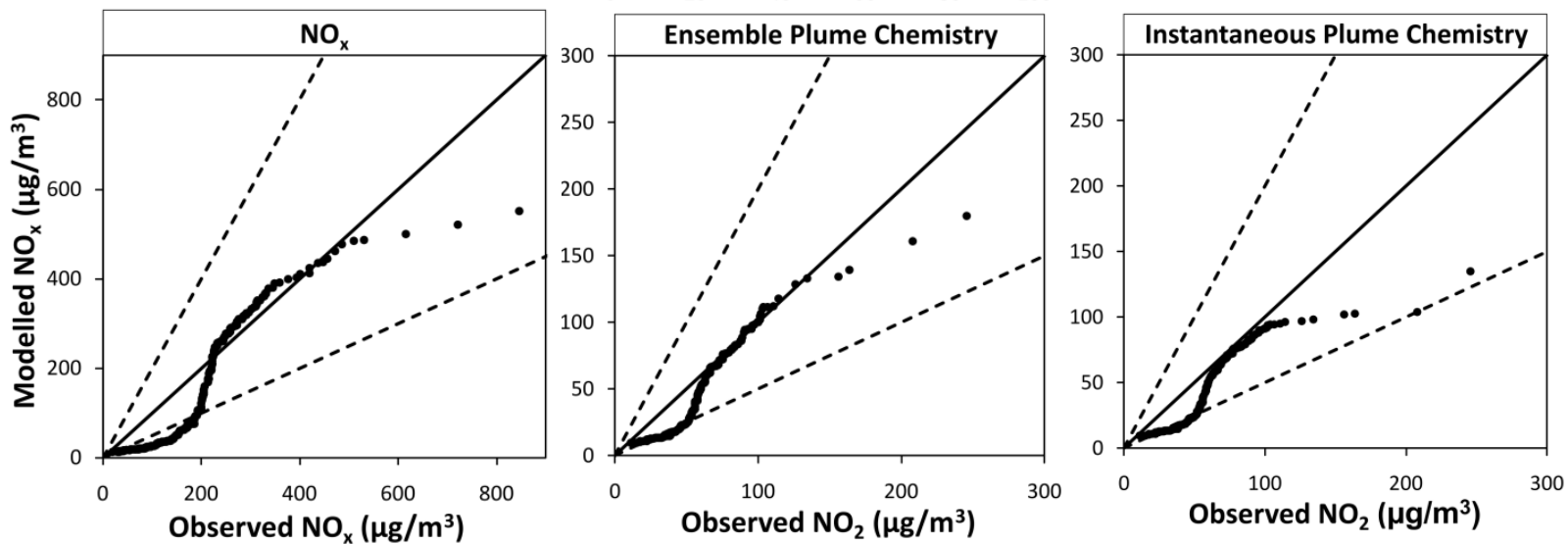
Observed Mean	190
Modelled Mean	160
R	0.68
Fac2	0.56
Observed Max	850
Modelled Max	550

## Case studies: Results

Wainwright

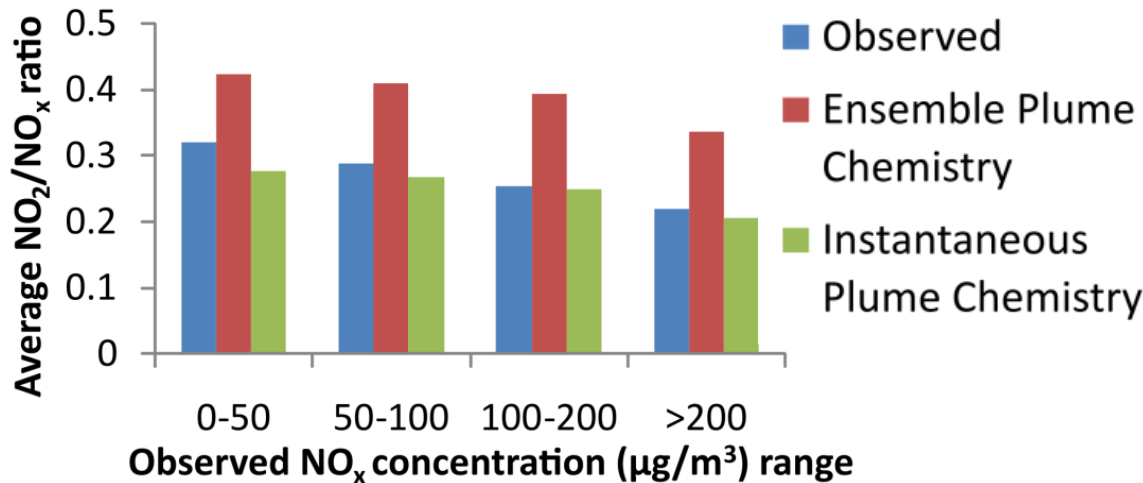


Prudhoe Bay

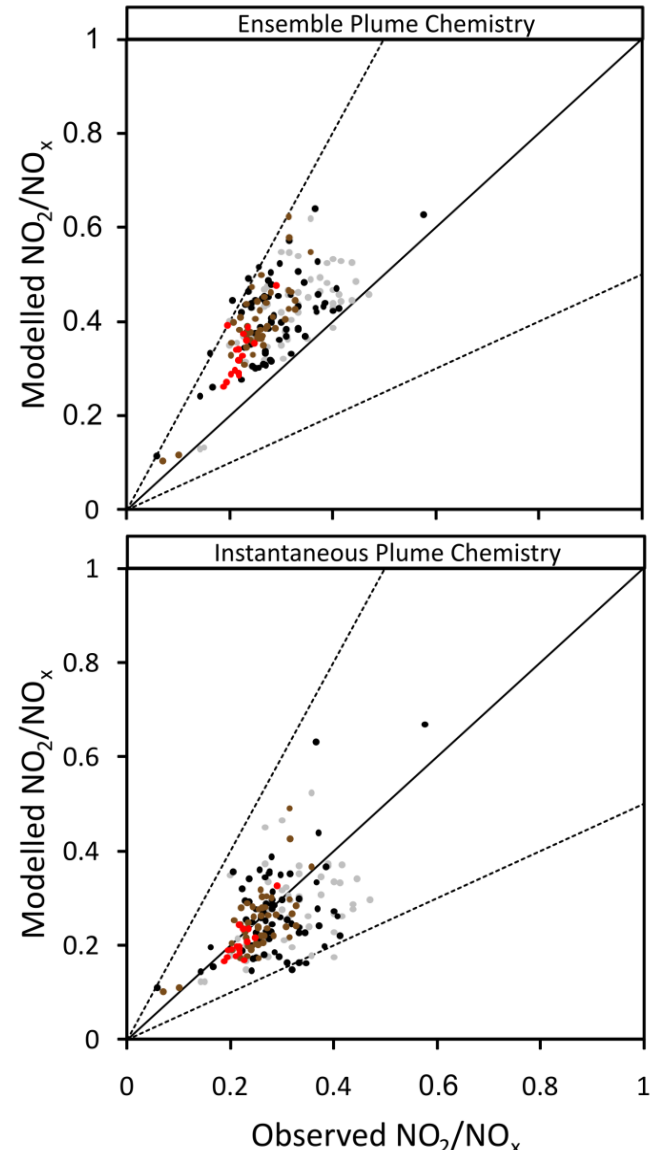
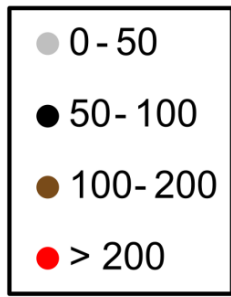


## Case study: Wainwright – NO<sub>2</sub>/NO<sub>x</sub> ratios

- Comparing NO<sub>2</sub>/NO<sub>x</sub> ratios gives good indication of chemistry performance
- Time-paired comparisons
- Instantaneous plume chemistry compares better with observed NO<sub>2</sub>/NO<sub>x</sub>



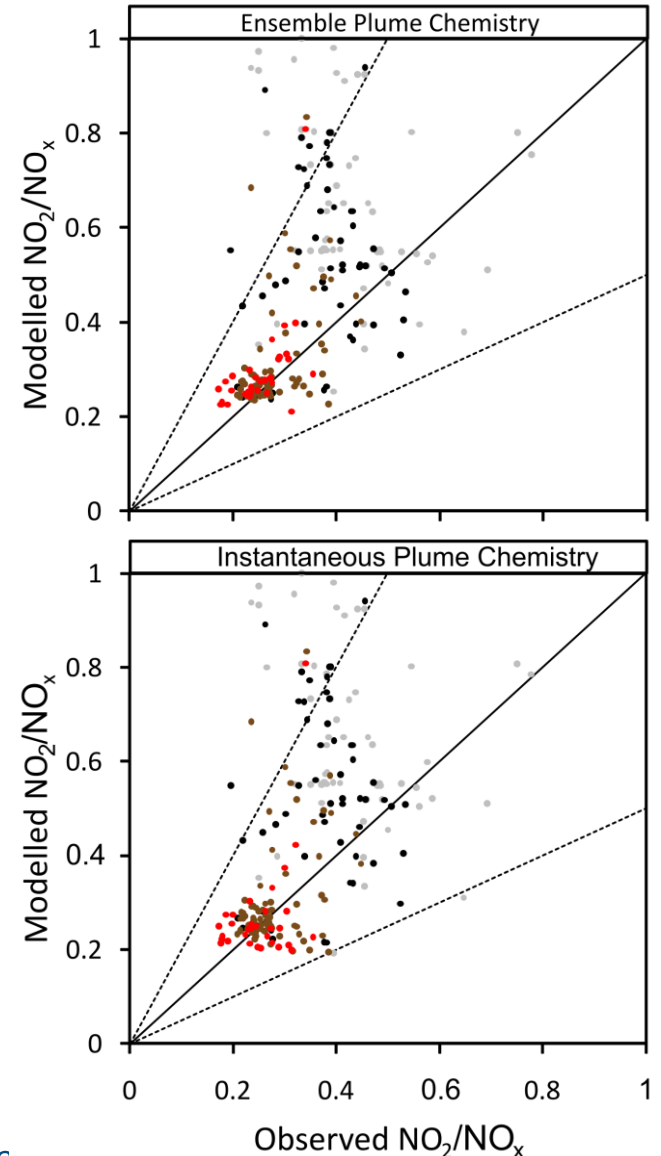
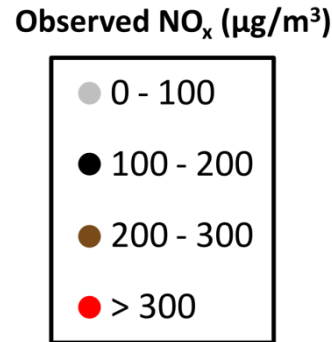
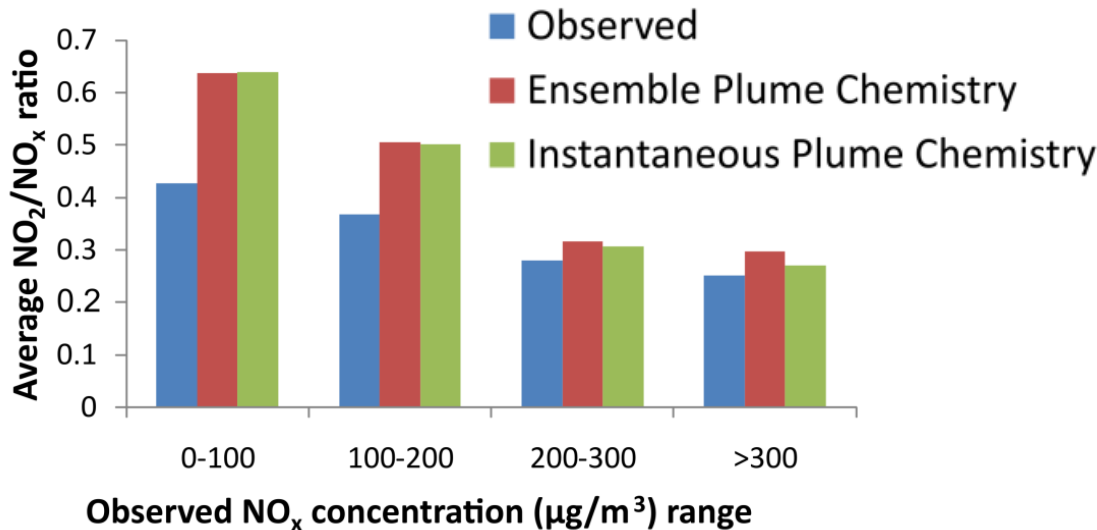
Observed NO<sub>x</sub> (µg/m<sup>3</sup>)



# Validation

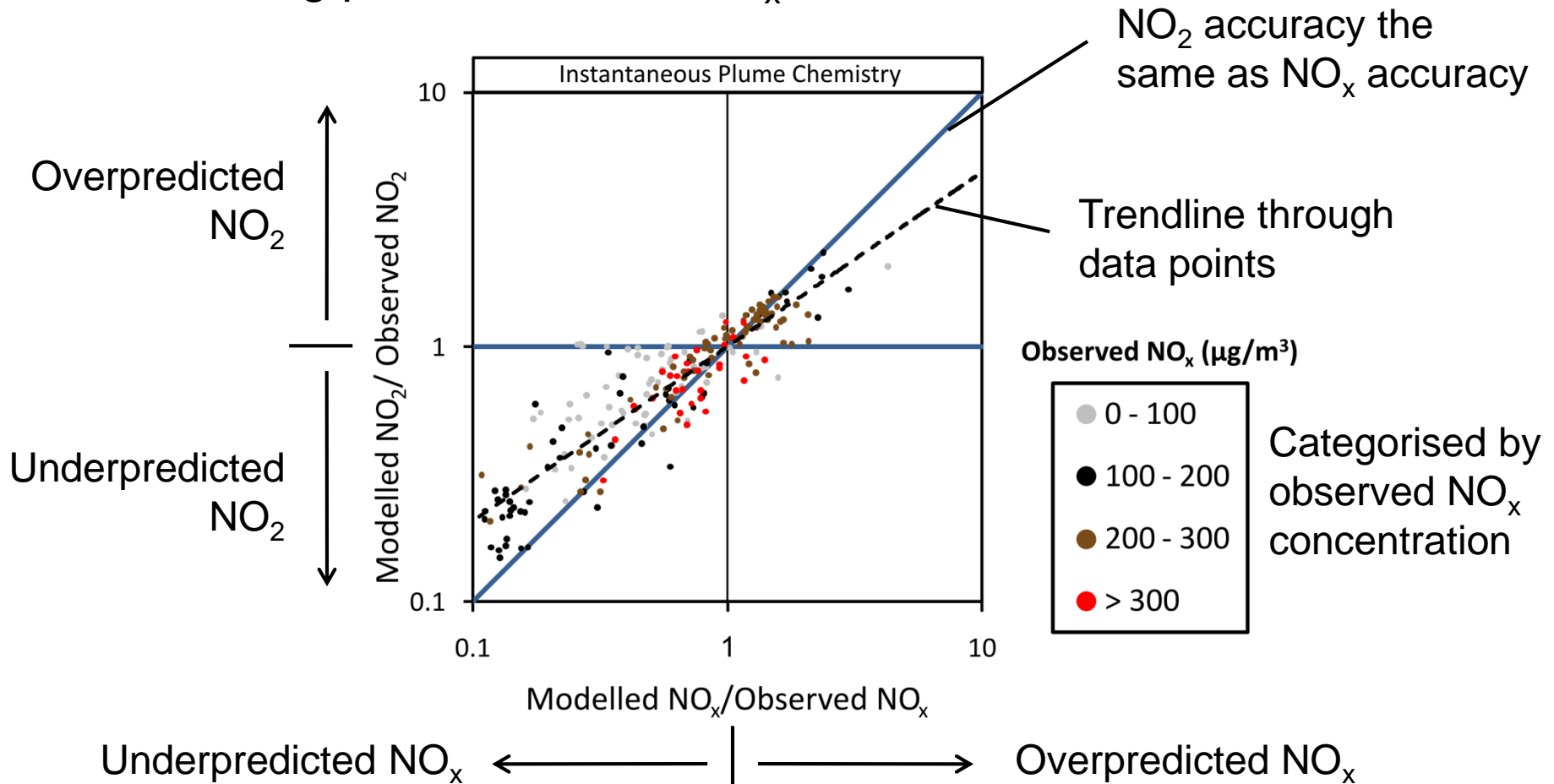
## Case study: Prudhoe Bay – NO<sub>2</sub>/NO<sub>x</sub> ratios

- Two chemistry schemes perform similarly
- Instantaneous plume chemistry predicts slightly more accurate NO<sub>2</sub>/NO<sub>x</sub>
- Better prediction for higher NO<sub>x</sub> concentrations



## Comparing modelled/observed ratios

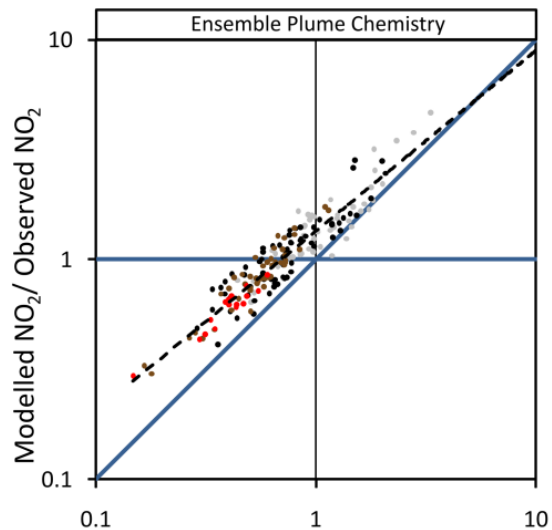
- Insightful to compare modelling performance of  $\text{NO}_2$  to modelling performance of  $\text{NO}_x$



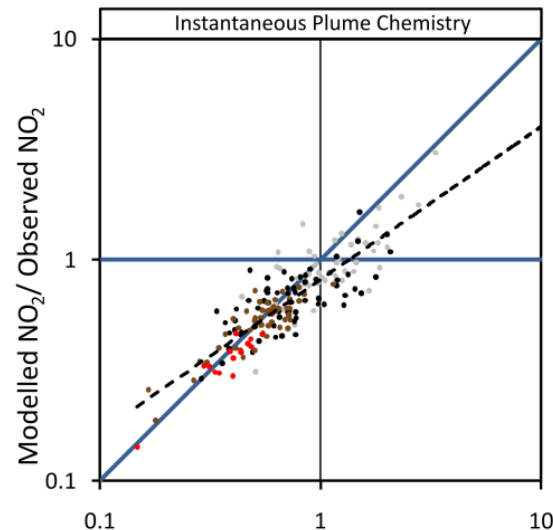
## Case studies: Results

Wainwright

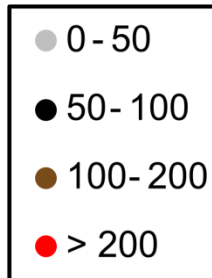
### Ensemble NO<sub>2</sub>



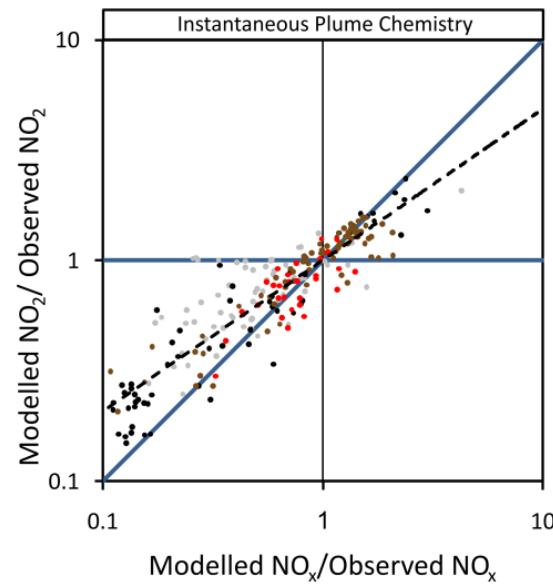
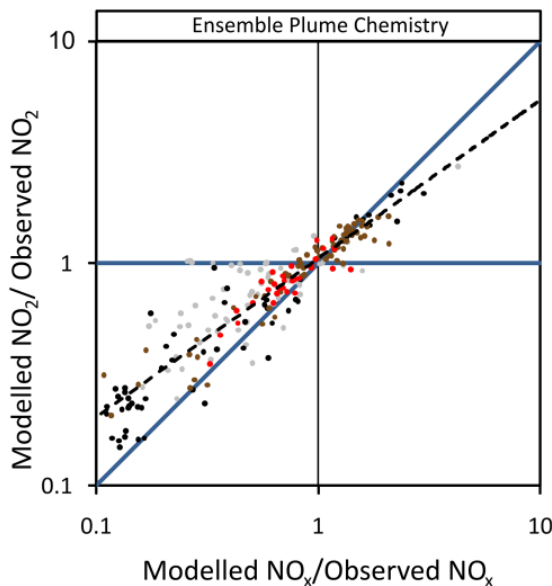
### Instantaneous NO<sub>2</sub>



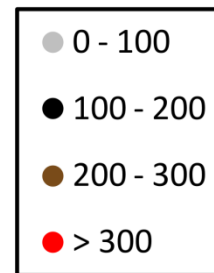
Observed NO<sub>x</sub> (μg/m<sup>3</sup>)



Prudhoe Bay



Observed NO<sub>x</sub> (μg/m<sup>3</sup>)





# Summary

- ADMS NO<sub>x</sub> chemistry validation for two schemes using two Alaskan case studies
- NO<sub>2</sub> model performance considered in the context of NO<sub>x</sub> model performance
  - Concentrations, NO<sub>2</sub>/NO<sub>x</sub> ratios and modelled/observed ratios were used to determine performance of chemistry models
- Both chemistry schemes compared well for both case studies
  - Ensemble plume chemistry predicted higher NO<sub>2</sub> concentrations than instantaneous plume chemistry
  - Instantaneous plume chemistry NO<sub>2</sub> concentrations compared better to measurements, when considered alongside modelled NO<sub>x</sub> performance

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**Thank you for listening**

**Any questions?**